

Synthesis of Acetamide Compounds from α -Pinene through Sonochemical Ritter Reaction with Ni/Natural Zeolite Catalyst

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Synthesis of Acetamide Compounds from α -Pinene through Sonochemical Ritter Reaction with Ni/Natural Zeolite Catalyst

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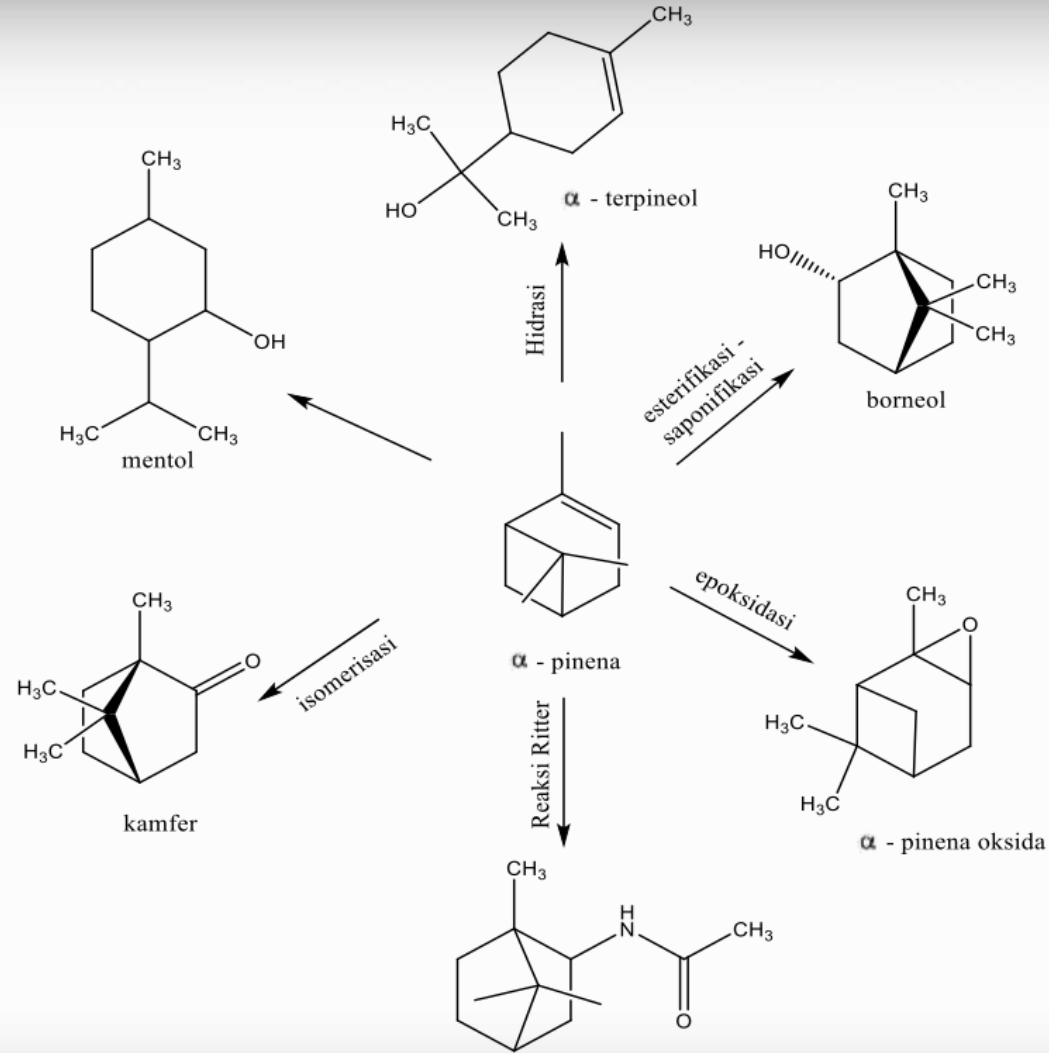
Malang, 14 Oktober 2022
International Conference on Organic and Applied Chemistry
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INTRODUCTION



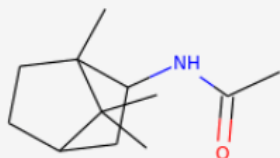
Terpentine oil

alpha pinene was modified into organonitrogen compounds through the Ritter reaction.



N-(1,7,7-trimethylbicyclo[2.2.1]heptan-2-yl)acetamide

Molecule Depiction



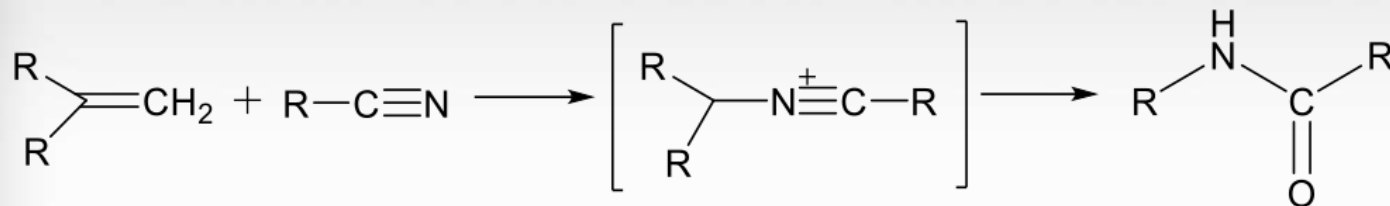
SMILES

Molecule properties:


Descriptor	Value
Molecular Weight	195.306
LogP	2.3373
#Rotatable Bonds	1
#Acceptors	1
#Donors	1
Surface Area	86.453

Property	Model Name	Predicted Value	Unit
Absorption	Water solubility	-2.996	Numeric (log mol/L)
Absorption	Caco2 permeability	1.638	Numeric (log Papp in 10 ⁻⁶ cm/s)
Absorption	Intestinal absorption (human)	94.681	Numeric (% Absorbed)
Absorption	Skin Permeability	-2.61	Numeric (log Kp)
Absorption	P-glycoprotein substrate	No	Categorical (Yes/No)
Absorption	P-glycoprotein I inhibitor	No	Categorical (Yes/No)
Absorption	P-glycoprotein II inhibitor	No	Categorical (Yes/No)
Distribution	VDss (human)	0.279	Numeric (log L/kg)
Distribution	Fraction unbound (human)	0.44	Numeric (Fu)
Distribution	BBB permeability	0.575	Numeric (log BB)
Distribution	CNS permeability	-2.207	Numeric (log PS)
Metabolism	CYP2D6 substrate	No	Categorical (Yes/No)

Ritter Reaction



ACID CATALYST

Ni/N-Zeolite  natural zeolite based heterogeneous catalyst with the addition of Ni

the addition of Ni metal can increase the acidity value which can increase its catalytic activity



The use of ultrasonic waves in the synthesis of chemical compounds encourages physical and chemical changes in liquid media and causes acoustic cavitation

OBJECTIVE

- The purpose of this study →
 1. to determine the reaction products obtained from the synthesis of acetamide compounds from α -pinene through the Ritter reaction sonochemically using Ni/Natural Zeolite catalysts
 2. To determine the effect of the type of catalyst for the synthesis of acetamide compounds from α -pinene through the sonochemical Ritter reaction

METHODS

Activation of
Natural Zeolite

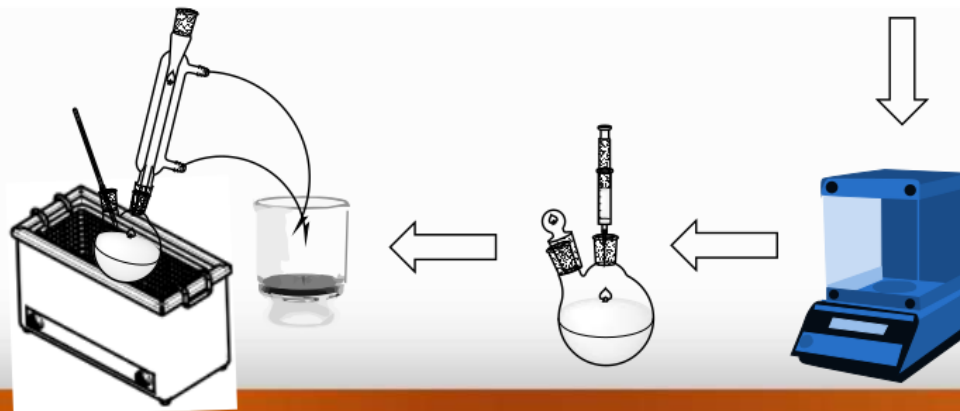
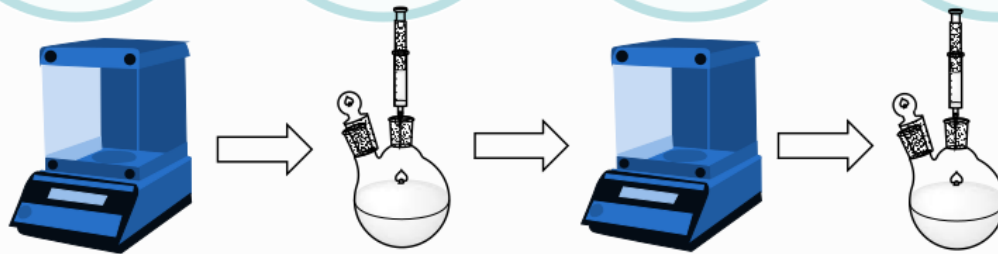
Impregnation of
Ni to Zeolite

Characterization
of catalyst

Synthesis
Acetamide
Compound

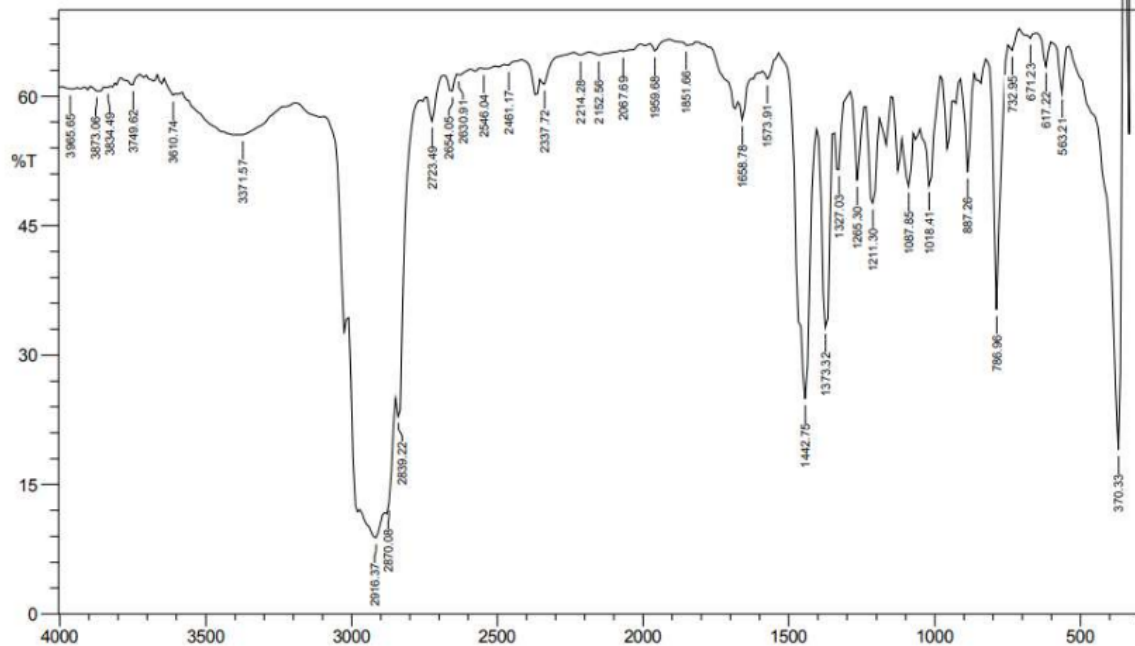
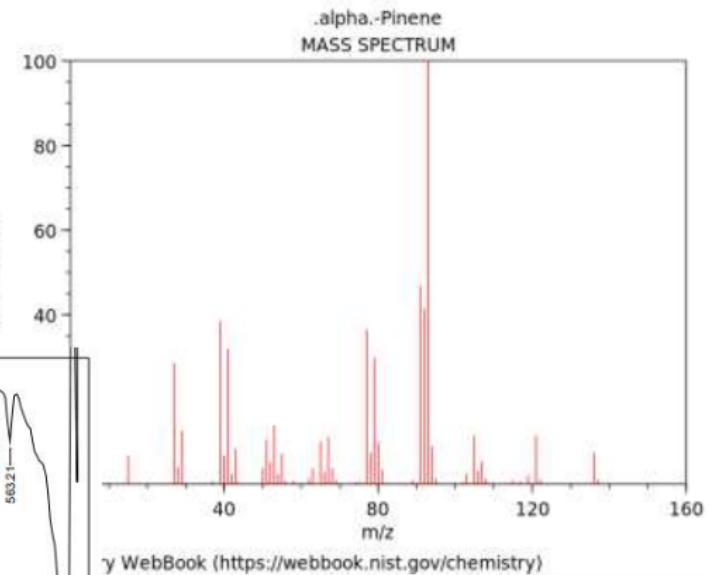
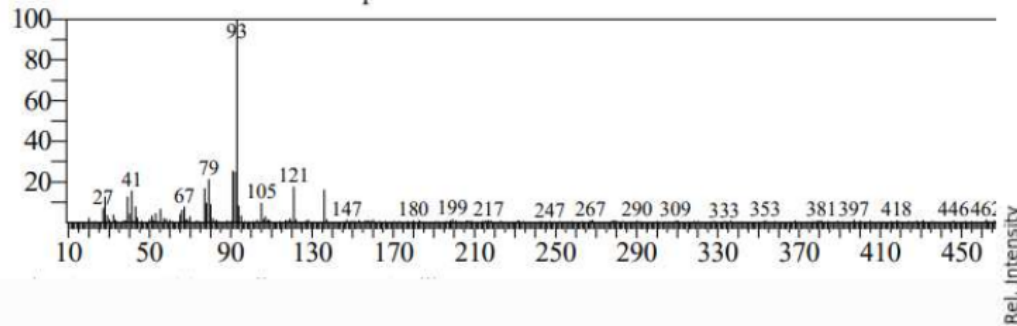
Separation of
product

Characterization
of product



RESULT AND DISCUSSION

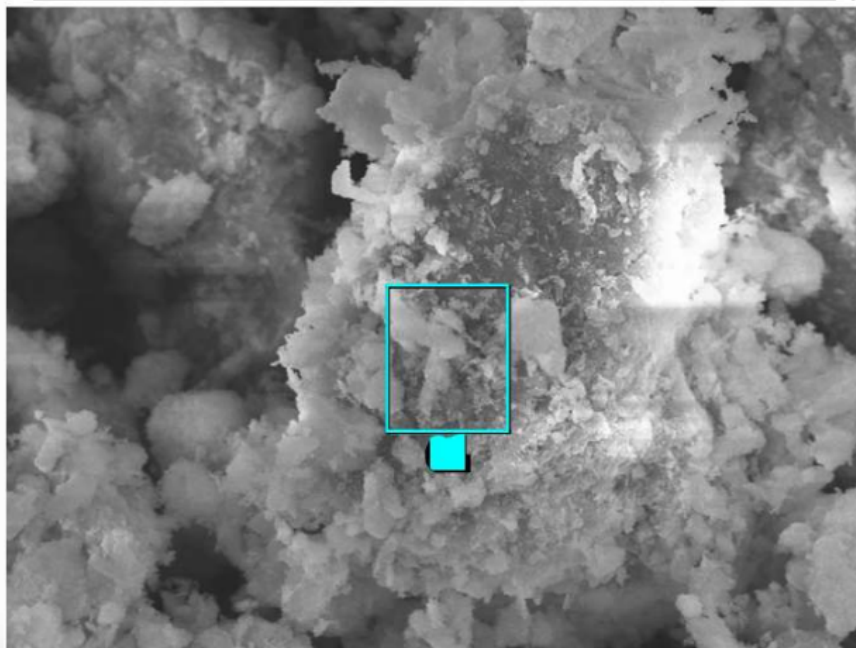
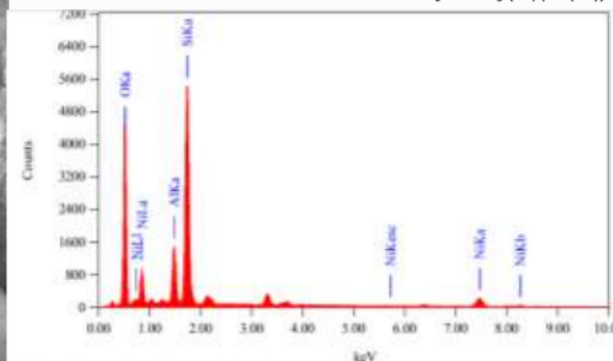
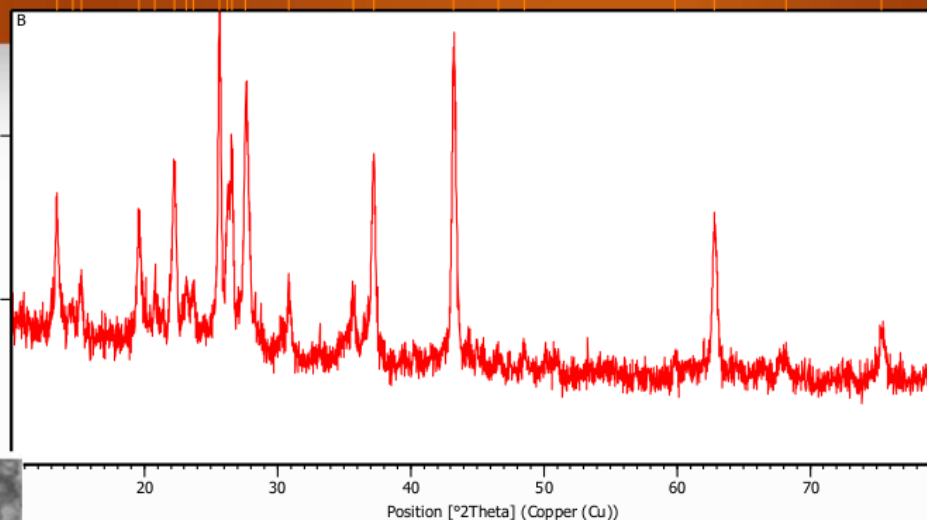
Alpha pinene as starting material



The levels of alpha-pinene in turpentine oil used as starting material is 93%

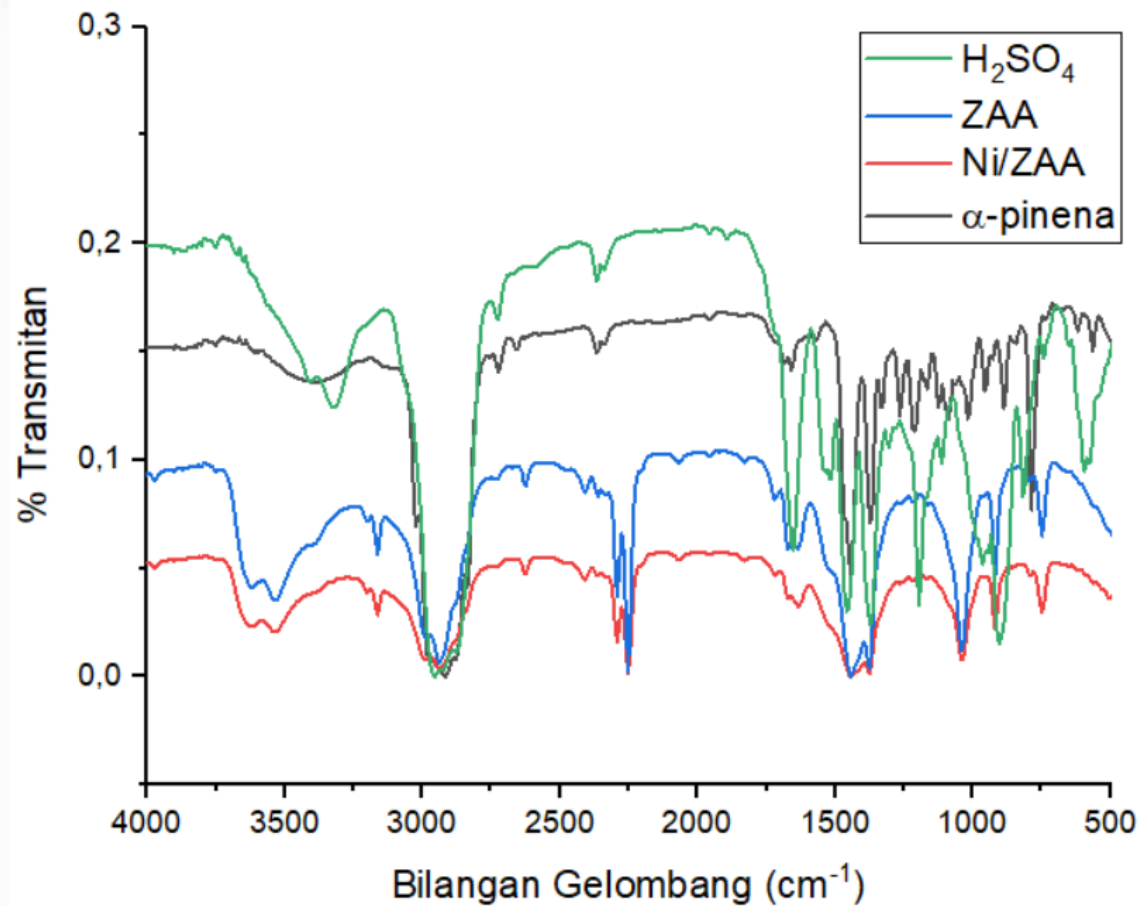
Charazterization of Ni/N-Zeolite Catalyst

	2θ	
	Standar	Ni/ZAA
Ni	35,840	35,6798
JCPDS No. 65-	37,80	37,2188
0380	50,146	42,2240
		62,7864

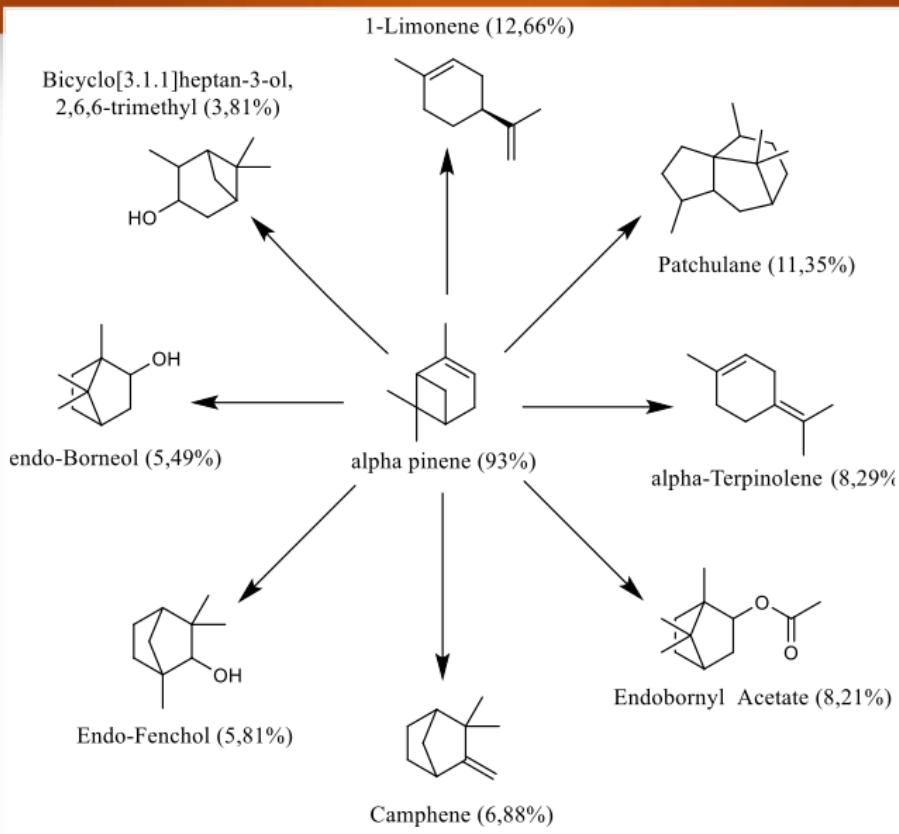


Elemen	Weight (%)
O	52,03
Al	6,87
Si	27,45
Ni	13,65
Total	100,0

Charazterization of Reaction Product with FT-IR



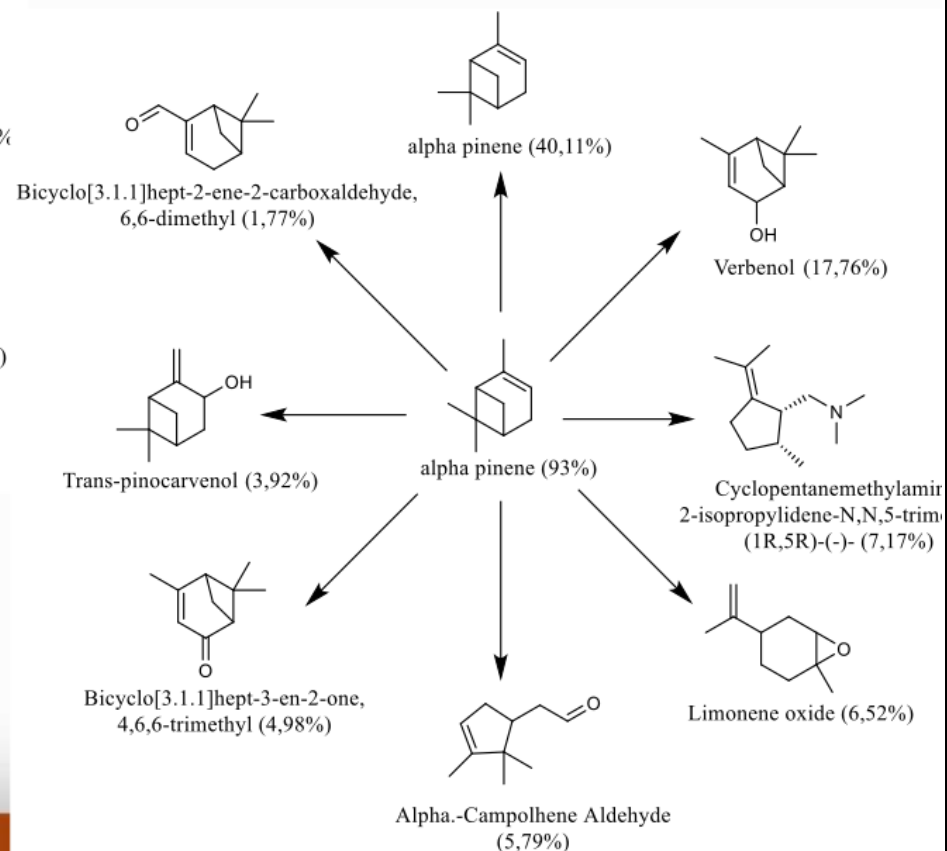
Charazterization of Reaction Product with GC-MS



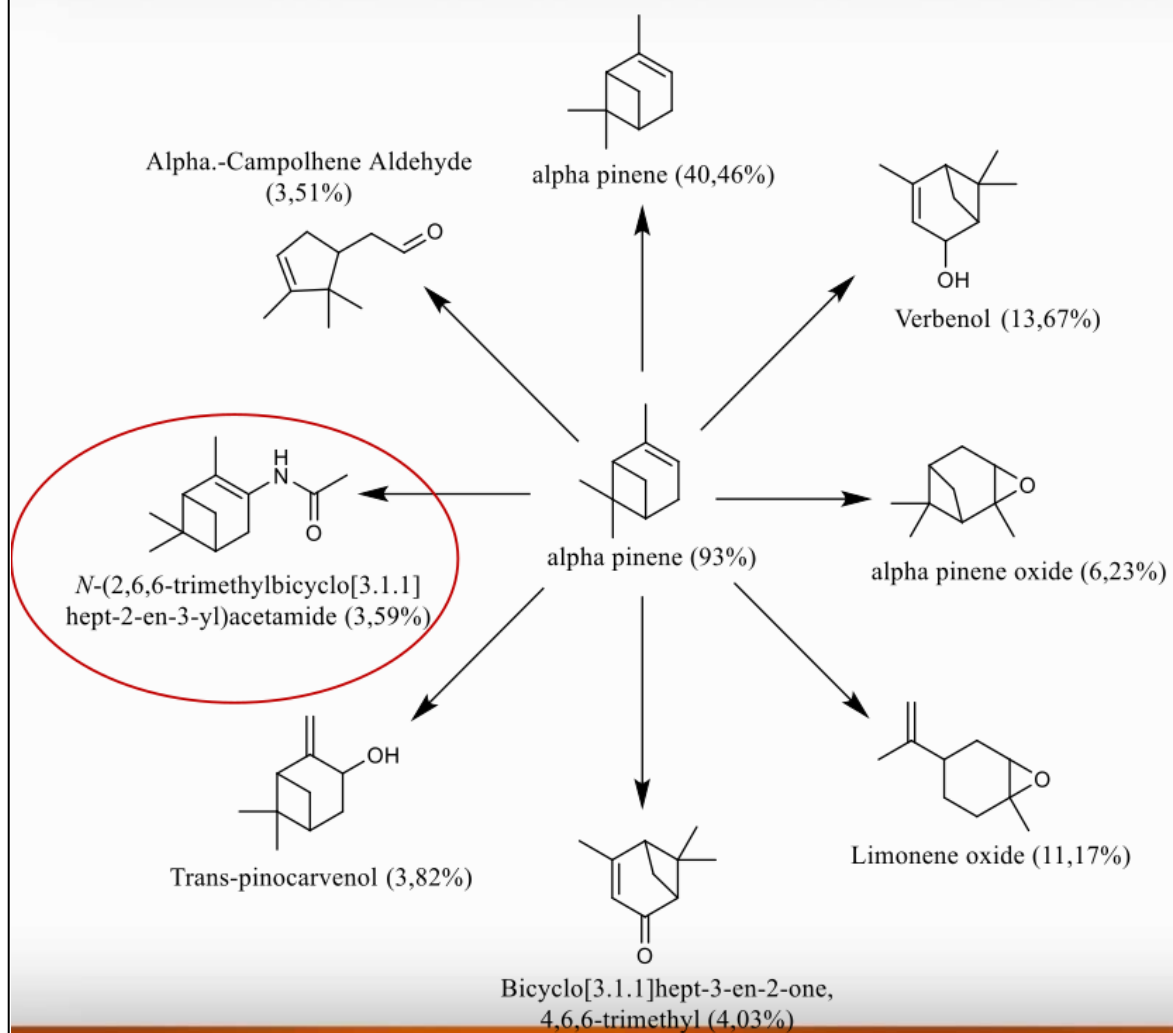
Reaction with natural zeolite catalyst by ultrasonication at room temperature for 5 h



Reaction with sulfuric acid by ultrasonication at room temperature for 5 h



Reaction with Ni/ZAA catalyst by ultrasonication at room temperature for 5 h



Conversion of alpha pinene = 54,56%
% yield of product = 4.1%

CONCLUSION

- Amide compounds are only formed in reaction products with a Ni/ZAA catalyst. Based on this, it can be concluded that the acidity of the catalyst affects the reaction products formed in the Ritter reaction. In addition, there is an effect of reaction time, in further research it is necessary to vary the reaction time using a Ni/Natural Zeolite catalyst

THANK YOU

Synthesis of Acetamide Compounds from α -Pinene through Sonochemical Ritter Reaction with Ni/Natural Zeolite Catalyst

GRADEMARK REPORT

FINAL GRADE

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GENERAL COMMENTS

Instructor

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